Introduction To Drug Aspirin and Its Impurity

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Abstract- Aspirin is one of the most frequently used and cheapest drugs in medicine. It belongs to the non-steroidal anti-inflammatory drugs with a wide range of pharmacological activities, including analgesic, antipyretic, and antiplatelet properties. Currently, it is accepted to prescribe a low dose of aspirin to pregnant women who are at high risk of preeclampsia (PE) because it reduces the onset of this complication. Drug produce degradation profiles essential to establish to monitor the stable formulation and provide appropriate drug shelf life valuation. Structural description of impurities and degeneracy production in bulk API has become integral part of pharmaceutical product development. The study of these minor leveled unidentified impurities and degradent are very challenging. Various regulatory bodies related International Council for Harmonisation, United States Food and Drug Administration.

Keyword: Aspirin, Drug, Impurity, Degradation, ASA, Chemical Composition.

Introduction  
Aspirin / acetylsalicylic acid (ASA) is a medicine used to lessen torment, fever, or irritation. Anti-inflammatory medicine was first disconnected Felix Hoffmann, a physicist was the German organization Bayer in 1897. Various medications that are accessible in market today were found from common sources. A significant model is the ibuprofen, which shows pain relieving movement. It is so far the world's most popular and most all around utilized therapeutic operator. Its source is from the plant genera Salix spp. also, Populus spp. what's more, it is identified with salicin.[1]

Chemical Composition  
Structural Formula - OCOCH3

Molecular Formula – C9 H8 O4  
Molecular Weight -180.00

Impurity Profile  
Impurity profiling is the universal term of a class of analytical activities, the purpose of which is the exposure, identification/structure explanation and measurable resolution of organic and inorganic contaminations just as remaining solvents in bulk drugs and pharmaceutical formations.[2]

The highest relevant application in the drug identification is safety related, no more individual of the drug but further impurities also degraded products present in them. Impurities today in the drug may lead to cytotoxicity, carcinogenic or teratogenic effects. For diseases like hypertension or diabetes which are related to change in body physiology, patient for his rest of life is going to be on medication. Though the amount of impurities is very minute still prolong exposure to them may be hazardous. Hence identification and the check on presence of specific amount is must. Drug produce degradation profiles essential to establish to monitor the stable formulation and provide appropriate drug shelf life valuation. [3]Structural description of impurities and degeneracy production in bulk API has become integral part of pharmaceutical product development.

Importance of Impurity Profile Study  
Major parts for pharmaceutical industries are product property, safeness and potency of drug. Stability of the drug is a quality attribute, which is associated with drug substance or drug products on account of purity, strength, identity, safety, apparent, chemical, physical, microbiological change, and they affect on the biological performance of drug product. Any variations in quality attributes of drug product with time are mandatory and it is directly proportional to safeness and potency of the drug.

Forced Degradation Study  
Stress study is a degeneracy of new medicament substance and medication product at conditions more severe than forward conditions. Constrained debasement considers show the synthetic presentation of the particle which thusly helps in the advancement of formulation and package. As analysis of dosage forms under stability study is an important. [4]Forced degradation studies appearance the chemical performance of the atom which is turn use in the improvement of formulation. In addition, the regulative guidance was very broad and does not explain about the work of stress studies.
Constrained debase measurements gives information to support identification of thinkable degradant; degradation paths and vital stability of the medication molecule and validation of stability representing analytical processes. A draft guidance document recommends results of one-time stress studies should include in Phase III Investigational New Drugs. NDA enrolment needs data of information of stress study concentrates as constrained debase items, debasement response energy, structure, drug peak purity and mass balance etc. This forced degradation study offer data about stress study pathways of API, alone and in drug item, any conceivable polymorphic or enantiomeric substances and change between drug related debasement and excipient interferences.[5]

Review of Literature
Deepti Jain. et al., (2013) [6]explained an ongoing patterns in analytical points of view of debase and polluting influences profiling of drugs including API just as medication items during 2008-2012. It centers unmistakably around thorough update of different explanatory strategies including hyphenated methods for the recognizable proof and evaluation of edges of impurities and degradant in various pharmaceutical networks. M. Dendeni et al., (2012) [7]reported different administrative specialists like International Council for Harmonisation (ICH), United States Food and Drug Administration are worrying on the immaculatesness necessities and the unmistakable evidence of impurities in active pharmaceutical drugs. Capacity of the contamination influences is the route toward getting and surveying data that sets up organic wellbeing of an entity contamination; in this manner, uncovering the need and degree of debasement profiling of medications in drug research.

Nadia M. Mostafa et al., (2016)[8] were created two sensitive, accuracy, precise, stability indicating methods and approved for the assurance of Ivabradine HCl (Iva) within the sight of its corrosive degradade, in mass powder and in drug definitions. Stress study was performed utilizing 2N HCl. The degraded item was distinguished by infrared spectroscopy and mass spectrometry, and the track of degradation was represented.

Prakash Katakam et al., (2014)[9] depicted Debasement Profiling has turn into a noteworthy Phase of Pharmaceutical investigation where both spectroscopic and chromatographic strategies discover applications. The investigative technique should be Touchy, Specific, and Precise which will isolate and choose the contamination of enthusiasm at the level 0.1%. Momentum research reports an endorsed RP-HPLC strategy to recognize also, isolate Valacyclovir-associated debasements (Imp-E And Imp-G) using the Box-Behnken.

Rishi Ram Parajuli et al., (2018)[10] were depicted drug contamination is any undesirable concoction or material that remaining parts with API which may have emerged out of union or may be made during definition measure or subsequent to developing of API and its plan. The adequacy and well being of pharmaceutical items might be impact by even limited quantities of undesirable, unidentified and possibly contaminations and lingering solvents in drug subtleties. Palavai reddy et al., (2012)[11] were delineated basic, touchy, and exact HPLC technique for the impurities profiling of naproxen and Sumatriptan succinate tablets has been made, endorsed and used for the affirmation of impurities in business pharmaceutical items. The Impurity were very much isolated on a Water Spherisorb ODS-1 segment (250mm X 4.6mm, 5μm) away the ingredient program utilizing methanol, Acetonitrile and 0.05 M Phosphate buffer (pH 3.0) at a stream pace of 1.0 mL min-1 with detection wavelength at 225 nm.

Shwetali K. Churi et al., (2017)[12] were clarified analytical chemistry is the investigation of material as for its subjective and quantitative piece. So as to guarantee the wellbeing and viability of all medications all through all periods of its shelf life, considering its storage, distribution and use, every pharmaceutical product should be analytically checked to discover the impurity level in any pharmaceutical medication with the assistance of new techniques and procedures like UV Spectroscopy, IR Spectroscopy, Mass Spectrometry, NMR spectroscopy and HPLC Methods.

Jaimes Torres et al., (2014)[13] explained about synthetic reactions are recognized among listat and sibutramine at low temperatures, reaction among listat and carnitine isn’t completed because of the absence of chlorine molecules, Experimental responses and secure qualities are clarified by hypothetical figurations, Great arrangement among theoretical and experimental results is acquired and Its IR spectra are anticipated utilizing DFT estimation.

Objectives
- To Procure gift sample of Aspirin from pharmaceutical company.
- Design and development of degradation protocol for Aspirin & its tablet dosage form.
- To perform stress study of Aspirin & its tablet dosage form underdifferent conditions according to ICH rules.
- To separate out degradant & Identification of Degraded Product byusing HPLC.
- To perform characterization of impurities by using IR spectroscopy,TLC & GCMS.

Research Methodology
Research Methodology refers the discussion regarding the specific methods chosen and used in a research paper. This discussion also encompasses the theoretical concepts that further provide information about the methods selection and application. The current study is descriptive in nature and is based on secondary data gathered from a variety of sources, including books, education, and development, journals, scholarly articles, government publications, and printed and online reference materials.

Result and Discussion
IR Study of Degraded Product-
Infrared spectroscopy is remarkable analytical techniques which offer the chance of chemical structural identification. IR helps to identify the functional groups in the structure of compound. The degraded compound from the various degradation methods was scanned. The IR was obtained by using the equipment JASCO FTIR-410 at Raja ram Bapu college of pharmacy, Kasegaon.[14]

![IR instrument - JASCO FTIR-410](image)

Table 1. Physical and Chemical Tests for Aspirin

<table>
<thead>
<tr>
<th>Test</th>
<th>Observation</th>
<th>Inference</th>
</tr>
</thead>
<tbody>
<tr>
<td>Chemical evaluation</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Appearance of solution</td>
<td>Clear</td>
<td>Confirmed</td>
</tr>
<tr>
<td>Clarity of solution</td>
<td>Clear</td>
<td>Confirmed</td>
</tr>
<tr>
<td>Solubility</td>
<td>Freely soluble in Ethanol &amp; Methanol, soluble in chloroform &amp; sparingly soluble in ether, Slightly soluble in water.</td>
<td>Confirmed</td>
</tr>
<tr>
<td>Limit test for Arsenic</td>
<td>Yellow color</td>
<td>Arsenic might be present</td>
</tr>
<tr>
<td>Limit test for Chloride</td>
<td>Gives Opalescence</td>
<td>Chloride might be present</td>
</tr>
<tr>
<td>Limit test for Sulphates</td>
<td>Gives Turbidity</td>
<td>Sulphates might be present</td>
</tr>
<tr>
<td>Physical evaluation</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Melting Point</td>
<td>135ºC</td>
<td>Confirmed</td>
</tr>
</tbody>
</table>

Based on physical and chemical evaluation it was confirmed that the given sample is Aspirin.[15-16]
Fig. 2 Chromatogram representing Aspirin and its alkaline degradation products at 0.1 N NaOH.

<table>
<thead>
<tr>
<th>Name</th>
<th>Retention Time</th>
<th>Area</th>
<th>% Assay</th>
<th>% Degradation</th>
</tr>
</thead>
<tbody>
<tr>
<td>Bulk Aspirin Alkaline</td>
<td>8.212</td>
<td>126654</td>
<td>37.49%</td>
<td>62.51%</td>
</tr>
<tr>
<td>degradation</td>
<td></td>
<td></td>
<td></td>
<td></td>
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</tbody>
</table>
Fig. 3 Chromatogram representing Aspirin and its oxidative degradation products at 3 % H2O2.

<table>
<thead>
<tr>
<th>Name</th>
<th>Retention Time</th>
<th>Area</th>
<th>% Assay</th>
<th>% Degradation</th>
</tr>
</thead>
<tbody>
<tr>
<td>Bulk Aspirin oxidative degradation</td>
<td>8.097</td>
<td>242823</td>
<td>71.88 %</td>
<td>28.22 %</td>
</tr>
</tbody>
</table>

**Conclusion**

Aspirin is Anti-rheumatic, Antithrombotic and Anti-inflammatory medication. Aspirin's capacity to decrease the production of thromboxanes and prostaglandins is because of its irreversible inhibition of COX enzyme. In present work, for analysis of Aspirin absorbance maxima was discovered to be 276 nm. For stability study conditions were optimized so that degradation between 5-20% of Aspirin can be achieved as per the ICH requirement. In the present work a detailed study and systematic profiling of impurity and degradation behavior of Aspirin have been completed by various techniques like TLC, FT-IR and GCMS is pending. Stress studies were completed to facilitate the advancement of stability indicating assay method (SIAMs) and to gain a better understanding of stability of Finished Pharmaceutical Product & API, to identify and quantify degradation products and to build up degradation pathways of degradation products formed during stress degradation. Significant degradation products were confined based on degradation kinetic study and were characterized.

**REFERENCES:**


