

# A Review On Synthesis And Characterization Of Metal Sulphide And Graphene Derivatives Based Advanced Materials For Supercapacitors

<sup>1</sup>Ankita Yadav, <sup>2</sup>Divesh

<sup>1</sup>PhD scholar, <sup>2</sup>PhD scholar

<sup>1</sup>Department of science

<sup>1</sup>SGT University, Gurugram, India

**ABSTRACT:** Affordable and clean energy is one of the major sustainable development goals that can transform our world. Currently, Researchers are focusing on cheap carbon electrode materials to develop energy storage devices, including high energy density supercapacitors and Li-ion batteries. In this report, the prime focus has been given towards different type of natural carbon sources used for synthesis of graphene and carbon product/derivative towards the fabrication of supercapacitors with high electrochemical performance. The review also contains the recent status of the synthetic methods of such type of materials and their subsequent applications as electrodes in supercapacitor made out of those natural resources. Supercapacitors are attracting considerable research interest as high performance energy storage devices that can contribute to the rapid growth of low power electronics and high power military power. This article reviews recent advancement in SC technology with respect to charge storage mechanism, electrode materials, electrolytes, and their practical applications. The challenges and opportunities associated with the commercialization of SCs are also discussed.

**Keywords:** Supercapacitors, Conducting polymers, Metal oxides, Electrolytes, energy storage devices, coal derivatives, Graphene.

## Introduction

The need for the new material gives birth to new technology how economic viability for any new technology is the prime recruitment for its sustainability the graphene and natural carbon derivative in the form of oxides (graphene oxides GO), dots (graphene quantum dots), and tubes is an output of the same lead. In view of the development of future technologies the same need present world requires enormous amount of carbon nanomaterials with promising an excellent chemical physical mechanical and thermal properties the year 1991 is known for the invention of graphene A SP hybridised material with extraordinary properties which open a wide opportunity for research and industries towards the development of various applications such as battery fuel cells and electronic devices. Graphene is also known for its large conjugated structure which signifies the delocalization effects of PI electrons this effect makes a graphene Band Gap material and restrict its use's in optronics application there are derivates of graphene in derivatives varied in the structural and chemical manner from each other to enhance the properties of these derivatives is an important aspect for its uses in various applications which can be done by intricately derivatives with the polymer matrixes since the industrial revolution the social and economic prosperity of the nation has dependent on the massive consumption of fossil resources coal gas and oil as a readily accessible carbon shows the total the total world energy demand from fossil fuels was estimated to be 13.731 billion tones of oil equivalent as of 2012 and is expected to approach 18.30 BTOE t2035 the depletion and uneven distribution of natural resources has already caused economic problems Example price fluctuations and unbalanced supply chains resulting in issues in various domains such as energy generation and storage industrial operation and transportation moreover the worldwide consumption of fossil fuel has caused carbon accumulation in the natural cycle. CO<sub>2</sub> from anthropogenic activities is the major contributor to the greenhouse gases causing significant environmental changes. In line with those efforts many studies have already considered how to restrict the consumption of fossil resources as part of those litigation strategy the development of renewable energy sources sets as solar energy tidal energy geothermal energy goals has also been used extensively Expect fossil fuels most unable energies are supplied as electricity. At such there has been great demand for reliable technical platform for electrochemical storage, including batteries, fuels electrochemical super capacitors in particular, SCs in particular assemblies have grown more attention than batteries because of their fast tourist capability that is low discharge time Sc 1 to 10s vs lithium iron Battery 10 to 60 minute and enhanced cyclic stability SC greater 13,000 H vs battery greater than 500h. in this review the compatibility of other active and inactive component of super capacitors with the activated carbon derived from biomass and natural sources will also be optimised also, various components of the activated carbon based electric double layer capacitance And their types will be reviewed and the novel status of art and binder which exhibit superior electrochemical performance will be discussed. Many reports own various types of Boron carbon obtained from different sources have been found which are chemically activated and used as electrode material in super capacitor In this regard naturally available biomass can be very useful as source material for the synthesis of porous carbon. Porous carbon prepared from these materials has an upper hand over the other in terms of sustainability as they are obtained in an environmentally friendly way from renewable waste by mass as well as natural sources there are very few reports on developing of porous carbon materials from waste biomass and other natural sources to be used in EDLC., the emphasis of is also given on activated carbon obtained from biomass the biomass material shows can be broadly summarised in five significant categories that is plant derived, food derived, microorganism derived And lastly carbon materials derived from natural sources such as coal. The technological advancement of modern society and the escalating global economy have used a huge demand for energy storage devices. As the reserves offer sale fuels are adjustable and considering the hazardous impacts on environment and climate

due to greenhouse gases come it is necessary for us to switch fuel to renewable energy sources Sustainable development full group the well-known renewable sources can't be used for commercial and resistant residences purpose as disruption in supply can cause large loss in the electrochemical energy storage has been regarded as the most promising among various renewable energy storage technology due to high efficiency and flexibility In recent years several new technologies like Electro chemical energy storage system such as secondary metal-ion batteries, redox-flow batteries, and electrochemical supercapacitors.

## 1. Review of literature

### 2.1 Synthesis of graphene derivative from coal/Graphite

#### a) Importance of precursor

Precursor always play an important role in the synthesis of graphene derivatives first look the size say, and yield of the final product dependent on the structure and properties of the precursor full slope the chemical structure and functional group present in the precursor always fault to produce the final and base grief in derivatives in past In past Graphite was a major precursor to synthesis and obtain the graphene derivatives. However in last few years researchers have started working on a broad range of precursors from organic inorganic, two natural sources like coal and biomass to synthesise graphene carbon derivatives.

#### b) Graphite as a precursor

The synthesis of graphene derivative from Graphite is a trend followed by researchers since past decades. Graphite can also be called a multilayer structured graphene. Graphite as a precursor also got attention due to its unique chemical structure and behaviour. The Graphite itself is capable of acting as an oxidising and reducing agent. Another important reason to use Graphite as a precursor is it's inter collecting capability of chemical bonds over the vessel plants. When a molecule intercalates between other two molecules such process is not as intercalation the uniqueness of Graphite is that each layer can either accept or donate electrons to other intra collected molecules without any loss of carbon atoms. This interaction projects help to without any loss of carbon atoms the intercalation process helps to vary the interlayer spacing between layers which decreases the strength of the material. This entire process known as the exfoliation process let the researcher to obtain single layer graphene sheets or graphene derivatives from Graphite precursor.

### 2.2 Structure of Graphene and its derivatives

Graphene and its derivative savoury similar structure that from 2D materials however they have some small differences that causes different physicochemical properties. In this section the scripture of graphene and its two main derivatives has been discussed.

#### Graphene

As the main critical element in graphene and its derivative is carbon understanding of carbon and its pounding is important Is carbon atom has four valence electron that could be saved through covalent bond with other elements. These electrons could be hybridised in different forms common namely  $sp$   $sp^2$  &  $sp^3$ . The 2D structure of graphene is a result of sharing SP square hybrid rise orbitals of 1 carbon with three neighbouring carbon atoms.

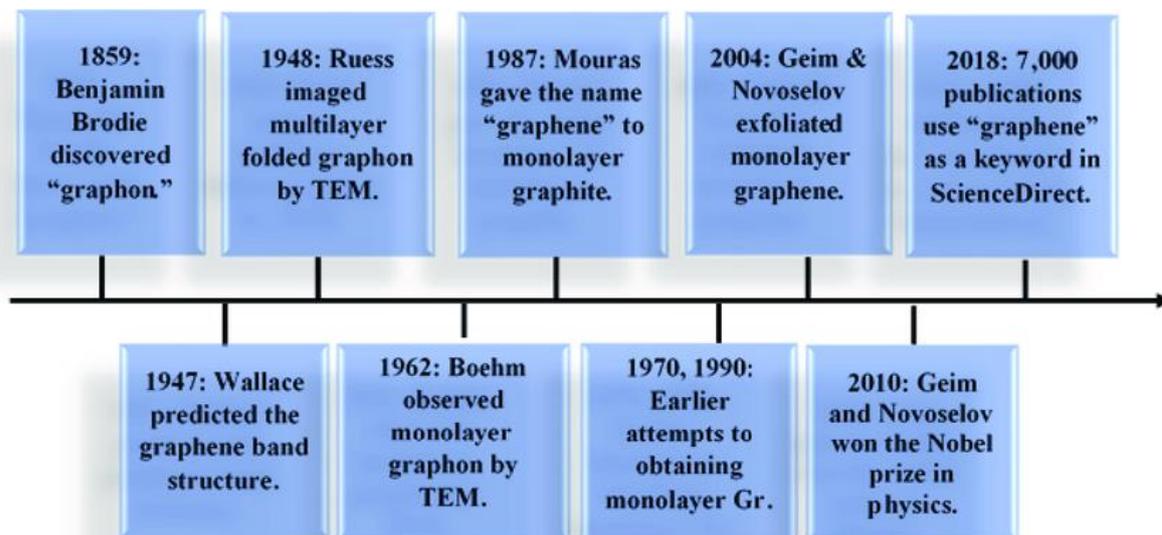


Fig.1: Key milestones in graphene derivatives

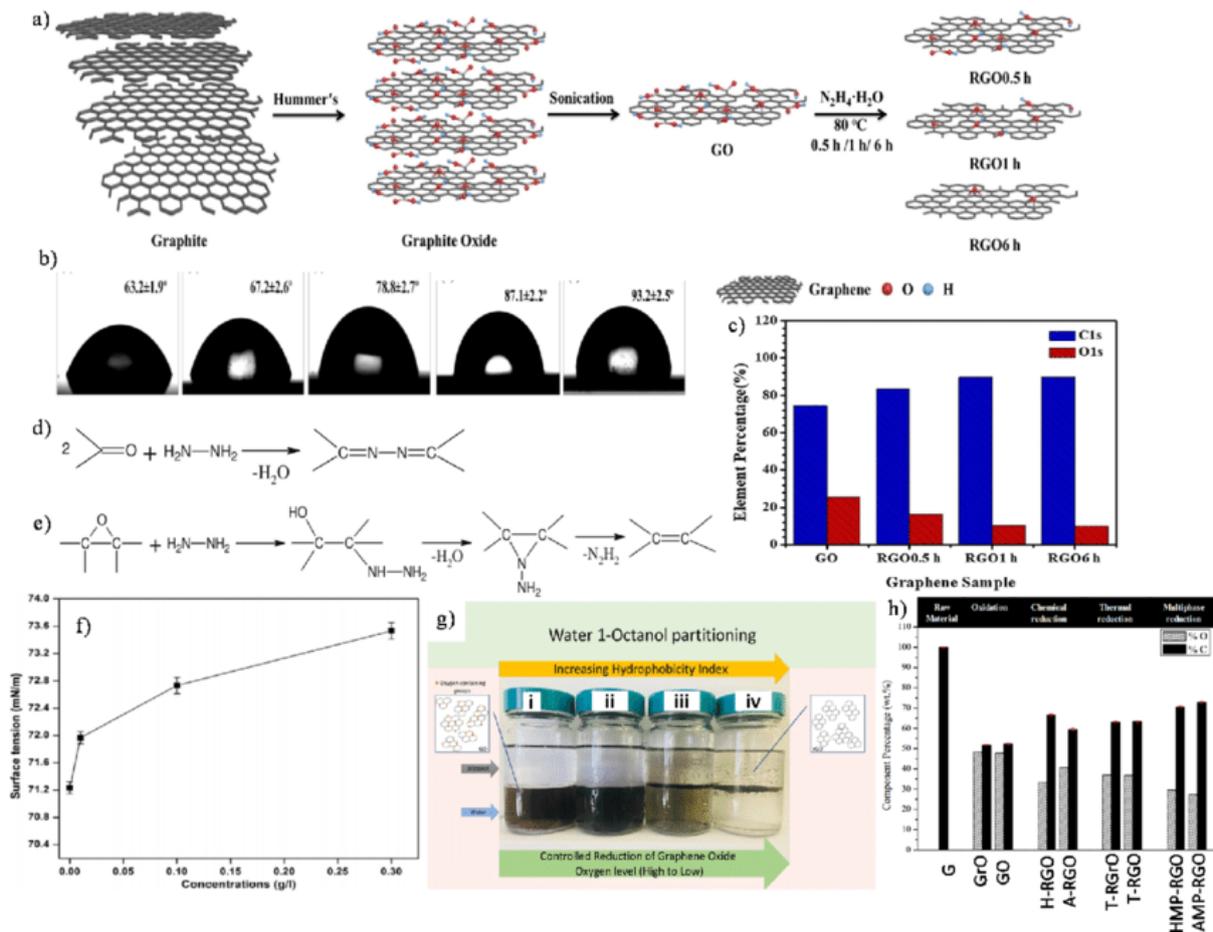


Fig. 2 : Graphene and its derivatives

**2.2.1 Synthesis and preparation of graphene**

Graphene can be produced by three major techniques x4 exfoliation chemical vapour deposition and chemical based techniques. One of Graphy’s critical feature is the number of layers of graphene sheets in order to achieve the desired properties graphene in sheets. The sheets must be separated from each other or else they are prone to aggregate to form new rudimentary graphite structures. The methods of synthesis promote the exfoliation of graphene sheets.

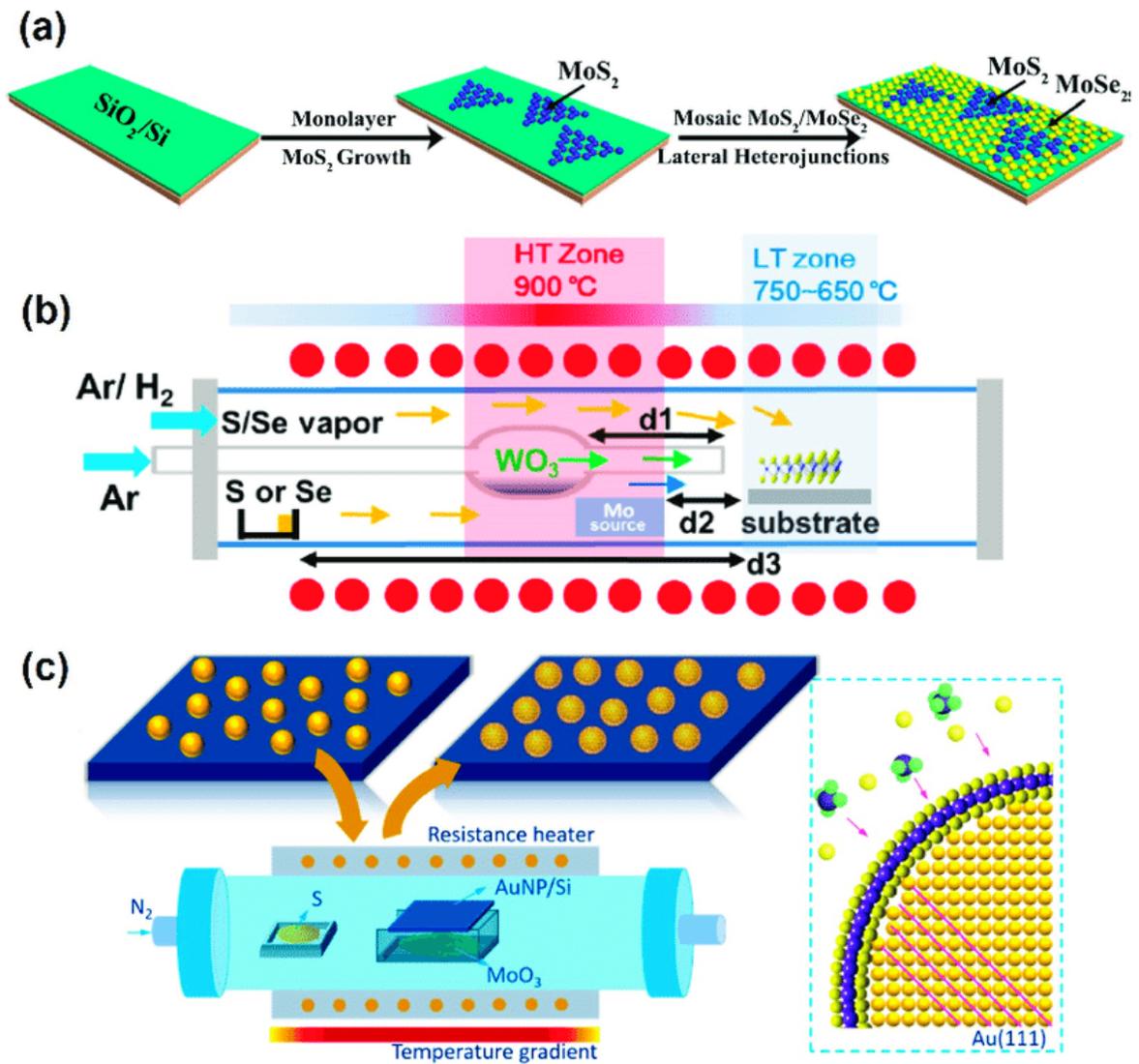
**a) Mechanical exploitation**

In order to harness the superior properties of graphene, sheets must be individually separated. Graphite is composed of Graphene seeds hailed together by van der Waals interaction And the method of extrusion takes advantage of this fact by breaking those weak interactions. Exfoliation is involved in other various techniques, including liquid phase exfoliation, exfoliation of Graphite to side un zipping carbon nanotubes, mechanical extrusion and more.

**b) Chemical vapour deposition**

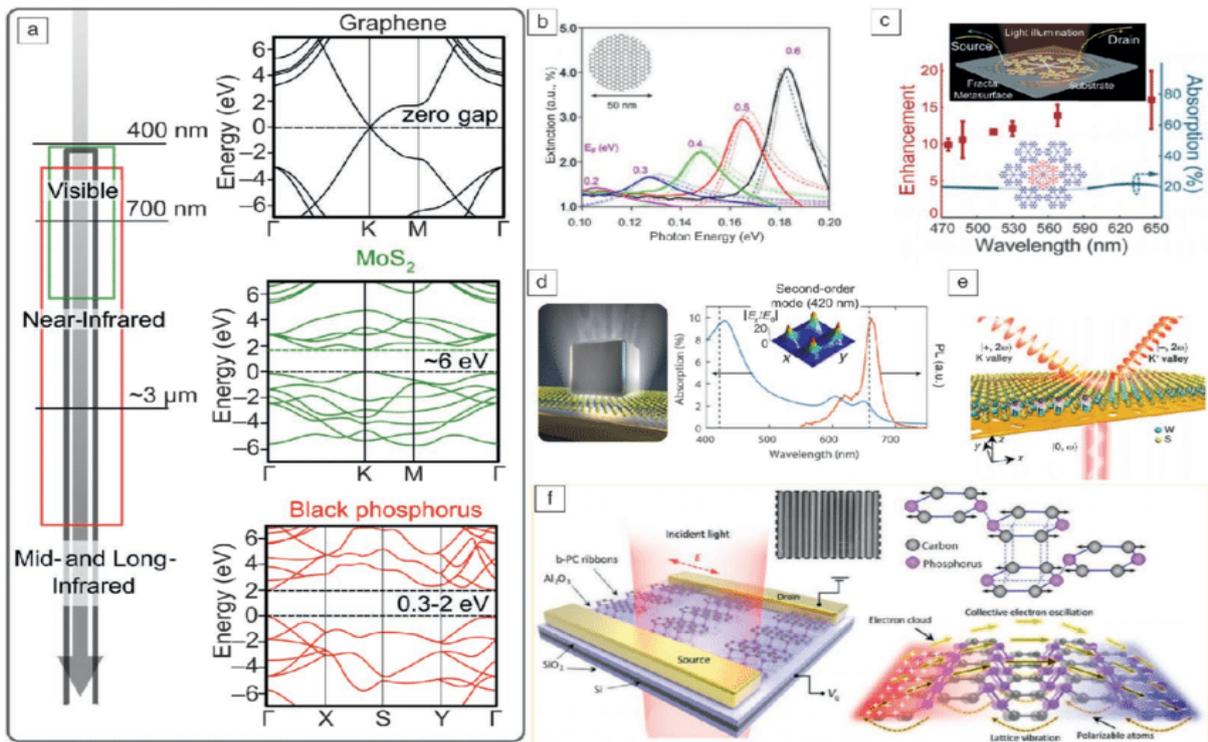
The chemical vapour deposition on transition metals example FECONICU is another method used to prepare graphene which requires a temperature as high as 1000 degree Celsius with a hydrocarbon gas flow as precursor this method is applied at high temperatures when exposing transition metal to low concentration hydrocarbon gas till the surface is saturated with carbon atoms Methane gas is journal used as source of carbon footstep hydrogen and organ are also used with methane as reaction stabiliser and enhancing the film uniformity.

Chemical paper deposition method (a) schematic illustration of the two steps CVD technique for the preparation of mosaic lateral heterojunctions film ohm SiO2substrate (b) symmetric is to illustration of the synthesis setup and various modes of hydro apitaxial growth of TMD atomic layers (c) symmetric solving the CVD process of MOS to sell growth on a Au nanoparticles.



**2.3 Recent progress on graphene research:**

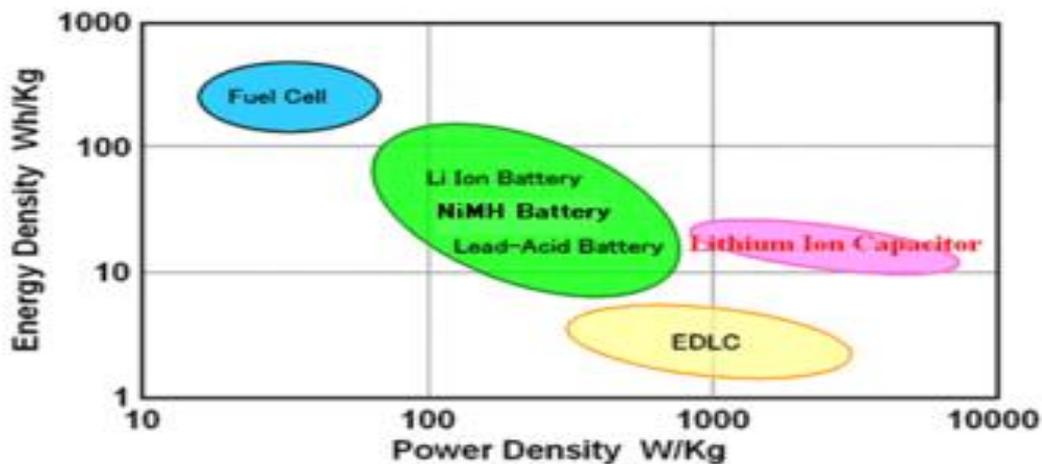
In recent years graphene based material have been extensively investigated for different applications. In this section the most recent application of graphene based material will be discussed in brief graphene has been extensively utilised for the development of electrode materials for super capacitors. for example graphene was combined with defect induced ITMOS 2 to develop a super capacitor electrode the hydrothermally synthesised hybrid electrode exhibited the specific capacitance of 442 F/g at the current density of 1A/g and cyclic stability of 90.3% upset 2000 cycles. In another study Elseway at al. synthesised and doped graphene from the waste polyethylene terephthalate bottles via a greens synthetic approach using urea as the end source.



(a) (Left) Band structures of single-layer graphene, MoS<sub>2</sub>, and black phosphorus. (Right) Spectral ranges covered by different materials (graphene [gray], MoS<sub>2</sub> [green], and black phosphorus [red]). (b) Extinction spectra of graphene nano disk array under different applied voltages, demonstrating large tunability. Inset shows a single graphene nano disk. Reprinted with permission from Reference 74. © 2019 American Chemical Society. (c) Measured photovoltage generation enhancement on fractal meta surface compared to plain source-drain contact. A broadband enhancement is achieved with a fractal meta surface and is insensitive to incident light polarization.

capacitance because of thinner dielectric and higher surface area electrodes these also allows poor **SUPERCAPACITORS: Construction and working of supercapacitors**

Conventional capacitors to stores energy by moving luck to one electoral to another before SC waste on carbon materials has a higher surface area, where phenomena known as the electric double layer is used to store charges. For SC involving metal oxide or polymeric material pseudo capacitors is the dominant charge storage mechanism. Don't supercapacitors and pill electrolyte capacitors are covered by the same capacitance equations SC can achieve higher power density greater than battery and energy density greater than capacitor.



specific area, availability, good conductivity, high electrochemical stability and wide operating temperature window. The (Ragon plot) above fig. shows the performance of high energy storage devices. SSC occupies space between batteries and capacitors this present a unique advantages that makes indispensable for applications which require high power delivered in a short time.

**SC fundamentals:**

Dielectric double layer phenomenon was first described by Helmholtz in 1853 and passenger by Becker in 1957 in material with high specific area as electrode four double layer structure formation the two electrodes are insulated by the membrane separator and impregnated to the electrolyte the membrane separate only permits the ion mobility but prevents electric contact SC Store Electric energy mainly through the formation of the double layer capacitors structure at the interface between the electrodes and the

electrolyte this energy studies mechanism involve no chemical freeze over composition changes apart irreversible parade reaction existing on the inner circle which also contribute to the total capacitance. The characteristics of electrostatic charge transfer results in a high degree of recyclability compared to conventional capacity Tapestors, the high capacitance of SC originate from the high specific area of the electrodes which is largely determined by the used deck rd material and their physical properties example conductivity and porosity. Advanced electrode material have been the area of intensive study and the last progress has been periodically reviewed in carbon materials with high specific area, conducting polymers and metal oxide constitute the main categories of SC electrode material Particularly carbon materials have been successfully utilised in the commercially available SC because of their advantages such as low cos,t high

#### SCs for the four application groups:

The most published topic is related to the use of SC in electricity traction applications this group represents 50% of the published studies. Most of them related to electric vehicles over hybrid electric vehicles (HEV).

The second group of the most published topic is related to power grid applications most of them are related to the improvement of the control strategy of a micro grid, the voltage and frequency regulation, and the increase of battery lifetime.

The third position is for the group of studies related to the renewable generation system, especially solar PV, wind and wave energy source is. Finally the last group with 10% of the papers are those applications related to the autonomous powers power systems, ships and aircrafts.

#### Conclusion:

Graphene derivatives based super capacitors will be the next generation energy storage device because of their promising attributes. This review gives the detailed explanation and reports on the synthesis of carbon/graphene derivatives using different precursors and methods of carbon/graphene derivatives providing the advantages of activated carbon derived from natural precursors for EDLC/Supercapacitors over other electrode material such as metal oxides CNTs conducting polymers, and other composites. There are reports of several noble as well as conventional methods to achieve high energy density of super capacitors. The compatibility of other inactive and active components with the activated carbon is observed to be having prominent role in fabrication of supercapacitors energy storage devices full stroke the various properties and novels synthetic methods of electrolytes, binders, current collectors, separators, and additives have been widely explored which leads to the development of super capacitors with excellent electrochemical properties. Form best of our literature survey the best conductive additive is found to be carbon black. Very few reports are available on the separators used in supercapacitors and this inactive material must be emphasised as wettability of the separator plays a vital role in performance supercapacitors apart from the commercially available organic electrolyte there are reports on ionic liquid tailored in novel wage for superior performance. Some of the key factors that must be considered before developing super capacitor energy storage devices are

1) the story's mechanism and other technology background must be studied in order to improve the chemical reaction occurring within the supercar.

2) the precursor of carbon glyph in derivative should be selected properly since the precursor containing high carbon content and inheritance interconnected mesoporous and micro porous network will serve as better and abundant materials Electrolyte, and binders for commercial applications.

All these factors and novel material and method must be researched in order to develop super capacitors to meet the energy demand in the near future.

Also in this paper some of the characteristics of the super capacitors have been discussed which will be helpful to selected super capacitor and design energy storage system using it with high power density so charging time large discharging time long life and environmentally friendly properties super capacitor may be chosen as an alternative for battery or other energy storage devices.

#### Future aspects:

Since graphene is an inspiration four other 2D materials, materials like hexagonal boron nitride, Germany, silicon, different transition battalion phosphorus and Boron nitride Nene seats are emerging as new and novel nanostructure for the next generation nanoscience and nanotechnology research. It is reported that in some cases these two de materials are quit better than graphene deep investigation on these new nano systems are currently underway with increasing achievements for prototype applications however more intense research and funding are required for their industrial scale applications. In this way these two de materials are new challenges for the materials scientist and a great opportunity for further research and development. Despite the usage of bulk graphene based nano bio material for novel bio applications few attempts have been done to realise cost effective scalable and reproducible synthesis methods of stable and reliable graph in based nanomaterials. In addition new biofunction methods have to be developed to prevent the graphene material from agglomeration during biomedical applications. Additionally graphene and its derivative have been introduced as a promising method in order to improve characteristics of dental materials. For example the bio composites with tuneable psychochemical biology biological properties that can be synthesised by functionalisation and combination of graphene and its derivatives with other biomolecules and biomaterials in order to obtain specific characteristics such as high mechanical properties, large surface area as well as enhanced bio activity Eventually we believe that the utilisation of graphene based engineered nanomaterials in the dentistry deserve to be profoundly examined as it can promote much progressively dependable dental treatments in the newer future.

Some of the fields were supercapacitors can be used are

- 1) in transmission lines SSCUPS for critical loads which need ride through a few second.
- 2) SC system with batteries are used hybrid SC battery UPSC and battery can compliment each other in their shortcoming which would reduce battery cycling in turn in increasing battery life system frequency and stability control.

## Acknowledgements

I wish to express deep sense of gratitude to supervisors Dr Gurjot Singh and Dr Mukesh Kumar Department of Physics, Faculty of Science, Shree Guru Gobind Singh Tricentenary University for their invaluable guidance and steady motivation throughout this review work. I am thankful to Dr Lakhwinder Singh, Dean, Faculty of Science, for timely support and encouragement. I would like to pay my emotions of respect and love to my Husband and parents, for their ever- loving support, and motivation, which gave me confidence and impetus to work hard to reach this destination.

**Ankita Yadav**

**References:**

1. S.K. Tiwari, V. Kumar, A. Huczko, R. Oraon, A.D. Adhikari, G. Nayak, Magical allotropes of carbon: prospects and applications, *Crit. Rev. Solid State Mater. Sci.* 41 (4) (2016) 257e317.
2. Y. Yang, R. Liu, J. Wu, X. Jiang, P. Cao, X. Hu, T. Pan, C. Qiu, J. Yang, Y. Song, D. Wu, Y. Su, Bottom-up fabrication of graphene on silicon/silica substrate via a facile soft-hard template approach, *Sci. Rep.* 5 (2015) 13480.
3. S.K. Tiwari, R.K. Mishra, S.K. Ha, A. Huczko, Evolution of graphene oxide and graphene: from imagination to industrialization, *Chem. Nano. Mat.* 4 (7) (2018) 598e620.
4. L. Liao, H. Peng, Z. Liu, Chemistry makes graphene beyond graphene, *J. Am. Chem. Soc.* 136 (35) (2014) 12194e12200.
5. Z.E. Hughes, T.R. Walsh, Computational chemistry for graphene-based energy applications: progress and challenges, *Nanoscale* 7 (16) (2015) 6883e6908.
6. G.W. Flynn, Atomic Scale Imaging of the Electronic Structure and Chemistry of Graphene and Its Precursors on Metal Surfaces, Columbia Univ., New York, NY (United States), 2015.
7. Y. Peng, Z. Wang, K. Zou, Friction and wear properties of different types of graphene nanosheets as effective solid lubricants, *Langmuir* 31 (28) (2015) 7782e7791.
8. J. Zhang, S. Li, B. Tang, Z. Wang, G. Ji, W. Huang, J. Wang, High photocatalytic performance of two types of graphene modified TiO<sub>2</sub> composite photocatalysts, *Nanoscale Res. Lett.* 12 (1) (2017) 457. [9] J.-H. Lee, E.K. Lee, W.-J. Joo, Y. Jang, B.-S. Kim, J.Y. Lim, S.-H. Choi, S.J. Ahn, J.R. Ahn, M.-H. Park, Wafer-scale growth of single-crystal monolayer graphene on reusable hydrogen-terminated germanium, *Science* 344 (6181) (2014) 286e289.
9. S.P. Surwade, S.N. Smirnov, I.V. Vlasiouk, R.R. Unocic, G.M. Veith, S. Dai, S.M. Mahurin, Water desalination using nanoporous single-layer graphene, *Nat. Nanotechnol.* 10 (5) (2015) 459.
10. Chukwuka C, Folly K. A, —Batteries and Super-capacitors, *IEEE PES PowerAfrica*, pp. 1 – 6, 2012.
11. A. Armutlulu, J. K. Kim, M. Kim, S. A. Bidstrup Allen, M. G. Allen, —Nickel-oxidebased supercapacitors with high aspect ratio concentric cylindrical electrodes, *Transducers & Eurosensors*, pp. 1480 – 1483, 2013.
12. C. D. Lokhande, D. P. Dubal, Oh-Shim Joo, —Metal oxide thin film based supercapacitors, *Current applied physics*, Vol.11, pp. 255 - 270, 2011.
13. J.Manikandan, R.MohanaPriya, S.Pavithra, D.Sarathkumar, —Rapid Smart Phone Charging Using SuperCapacitor, *International Journal of Advanced Research in Electrical, Electronics and Instrumentation Engineering*, Vol. 4, pp. 2175 - 2180, 2015.
14. Ionuț Ciocan, Cristian Fărcaș, Alin Grama, Adrian Tulbure, —An Improved Method for the Electrical Parameters Identification of a Simplified Pspice Supercapacitor Modell, *IEEE SIITME*, pp. 171 - 174, 2016
15. R. Kotz, M. Carlen, —Principles and applications of electrochemical capacitors, *Electrochimica Acta*, Vol. 45, pp. 2483 - 2498, 2000.
16. A.K. Geim, K.S. Novoselov, The rise of graphene, *Nanoscience and Technology: A Collection of Reviews From Nature Journals*, World Scientific, 2010, pp. 11–19.
17. M. Orlita, ., Approaching the Dirac point in high-mobility multilayer epitaxial graphene, *101(26)* (2008), p. 267601.
18. C. Lee, Measurement of the elastic properties and intrinsic strength of monolayer graphene, *321(5887)* (2008), pp. 385–388.
19. X. Huang, Graphene-based materials: synthesis, characterization, properties, and applications *7 (14)* (2011) 1876–1902.
20. A.C. Neto, F. Guinea, N.M.J.P.W. Peres, Drawing conclusions from graphene, *19 (11)* (2006) 33.
21. G. Eda, M.J.A.m. Chhowalla, Chemically derived graphene oxide: towards large area thin-film electronics and optoelectronics, *22(22)* (2010), pp. 2392–2415.
22. C.e.N.e.R. Rao, Graphene: the new two-dimensional nanomaterial, *48(42)* (2009), pp. 7752–7777.
23. Z.-S. Wu, Synthesis of high-quality graphene with a pre-determined number of layers, *47 (2)* (2009) 493–499.
24. J. Zhao, Efficient preparation of large-area graphene oxide sheets for transparent conductive films, *4 (9)* (2010) 5245–5252. [10] C.-G. Lee, et al., Integration of reduced graphene oxide into organic field-effect transistors as conducting electrodes and as a metal modification layer, *95(2)* (2009), p. 188.
25. M.J. McAllister, Single sheet functionalized graphene by oxidation and thermal expansion of graphite, *19 (18)* (2007) 4396–4404.
26. R. Peierls, Quelques proprietes typiques des corps solides. *Ann. IH Poincare*, 5 (1935), pp. 177–222.
27. L. Landau, Zur Theorie der phasenumwandlungen II, *Phys. Z. Sowjetunion* 11 (545) (1937) 26–35.
28. N.D. Mermin, Crystalline order in two dimensions, *Phys. Rev.* 176 (1) (1968) 250. [15] J. Venables, G. Spiller, M. Hanbucken, Nucleation and growth of thin films, *Rep. Prog. Phys.* 47 (4) (1984) 399.
29. J. Evans, P. Thiel, M.C. Bartelt, Morphological evolution during epitaxial thin film growth: formation of 2D islands and 3D mounds, *Surf. Sci. Rep.* 61 (1–2) (2006) 1–128.
30. Dreyer DR, Park S, Bielawski CW, Ruoff RS. The chemistry of graphene oxide. *Chem Soc Rev* 2010;39:228–40. <https://doi.org/10.1039/B917103G>.

31. Novoselov KS, Geim AK, Morozov SV, Jiang D, Zhang Y, Dubonos SV, et al. Electric field effect in atomically thin carbon films. *Science* 2004;306(5696):666–9.
32. Zheng J, Zhang H, Dong S, Liu Y, Nai CT, Shin HS, et al. High yield exfoliation of two-dimensional chalcogenides using sodium naphthalenide. *Nat Commun* 2014;5:2995–3002. <https://doi.org/10.1038/ncomms3995>.
33. Zhu Y, Murali S, Cai W, Li X, Suk JW, Potts JR, et al. Graphene and graphene oxide: synthesis, properties, and applications. *Adv Mater* 2010;22(35):3906–24. <https://doi.org/10.1002/adma.201001068>.
34. Zhang T, Liu J, Wang C, Leng X, Xiao Y, Fu L. Synthesis of graphene and related two-dimensional materials for bioelectronics devices. *Biosens Bioelectron* 2017;89:28–42. <https://doi.org/10.1016/j.bios.2016.06.072>.
35. Pal SK. Versatile photoluminescence from graphene and its derivatives. *Carbon* 2015;88:86–112. <https://doi.org/10.1016/j.carbon.2015.02.035>.
36. Ma Z, Dou S, Shen A, Tao L, Dai L, Wang S. Sulfur-doped graphene derived from cycled lithium-sulfur batteries as a metal-free electrocatalyst for the oxygen reduction reaction. *Angew Chem Int Ed Engl*. 2015;54(6):1888–92. <https://doi.org/10.1002/anie.201410258>.
37. Meric I, Han MY, Young AF, Ozyilmaz B, Kim P, Shepard KL. Current saturation in zero-bandgap, top-gated graphene field-effect transistors. *Nat Nanotechnol* 2008;3(11):654–9. <https://doi.org/10.1038/nnano.2008.268>.
38. Obata S. High crystallinity graphene synthesis from graphene oxide. *Carbon* 2017;114:750. <https://doi.org/10.1016/j.carbon.2016.11.043>.
39. Jeong HK, Lee YP, Lahaye RJWE, Park MH, An KH, Kim IJ, et al. Evidence of graphitic AB stacking order of graphite oxides. *J Am Chem Soc* 2008;130(4):1362–6. <https://doi.org/10.1021/ja076473o>. [41] Kucinskis G, Bajars G, Kleperis J. Graphene in lithium ion battery cathode materials: a review. *J Power Sources* 2013;240:66–79. <https://doi.org/10.1016/j.jpowsour.2013.03.160>.
40. Su FY, He YB, Li B, Chen XC, You CH, Wei W. Could graphene construct an effective conducting network in a high-power lithium ion battery? *Nano Energy* 2012;1(3):429–39. <https://doi.org/10.1016/j.nanoen.2012.02.004>.
41. Wissler M. Graphite and carbon powders for electrochemical applications. *J Power Sources* 2006;156(2):142–50. <https://doi.org/10.1016/j.jpowsour.2006.02.064>.
42. Khutia M, Joshi GM, Deshmukh K, Pandey M. Preparation of modified polymer blend and electrical performance. *Compos Interfaces* 2015;22(3):167–78. <https://doi.org/10.1080/15685543.2015.999215>.
43. Asghar A, Samad YA, Lalia BS, Hashaikeh R. PEG based quasi-solid polymer electrolyte: mechanically supported by networked cellulose. *J Membr Sci* 2012;421:85–90. <https://doi.org/10.1016/j.memsci.2012.06.037>.